

InfoSafe No™ VAR8J

Issue Date : November 2024

ISSUED by MILESTON

Product Name **LEMON BLEACH**

Section 1 - Identification

Product Identifier LEMON BLEACH
Company Name Milestone Chemicals Pty. Ltd. (ABN 85115166357)
Address 115 Northern Road West Heidelberg
VIC 3081 AUSTRALIA
Telephone/Fax Number Tel: (03) 9450 4555
Fax: (03) 9457 5518
Emergency Phone Number (03) 9450 4555 Mon-Fri 8am - 6pm
Recommended use of the chemical and restrictions on use As a bleaching and sanitising agent.

Section 2 - Hazard(s) Identification

GHS Classification of the Substance/Mixture Eye damage/irritation: Category 2A
Skin corrosion/irritation: Category 2
Signal Word WARNING
Hazard Statement (s) H315 Causes skin irritation.
H319 Causes serious eye irritation.
Pictogram (s) Exclamation mark



Precautionary Statement – Prevention P264 Wash contaminated skin thoroughly after handling.
P280(f) Wear protective gloves/protective clothing/eye protection/face protection.
Precautionary Statement – Response P302+P352 IF ON SKIN: Wash with plenty of water.
P305+P351+P338 IF IN EYES: Rinse cautiously with water for several minutes.
Remove contact lenses, if present and easy to do. Continue rinsing.
P332+P313 If skin irritation occurs: Get medical advice/attention.
P337+P313 If eye irritation persists: Get medical advice/attention.
P362+P364 Take off contaminated clothing and wash it before reuse.
Precautionary Statement – Storage P404 Store in a closed container.
Precautionary Statement – Disposal P501 Dispose of contents/container: Recycle packaging by replacing cap and returning clean containers to recycler or designated collection point.
Precautionary Statement – General P102 Keep out of reach of children.
P103 Read carefully and follow all instructions.

Section 3 - Composition and Information on Ingredients

Ingredients	Name	CAS	Proportion
	Water	7732-18-5	60-100 %
	Other ingredients determined not to be hazardous	Not required	0-10 %
	Sodium hypochlorite	7681-52-9	0-5 %

Section 4 - First Aid Measures

Inhalation Remove from exposure, rest and keep warm. In severe cases, obtain medical attention.
Ingestion Immediately rinse mouth with water. Do NOT induce vomiting. Slowly give water to drink. Seek medical assistance.

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Skin	If skin contact occurs, remove contaminated clothing and wash skin thoroughly. If swelling, redness, blistering, or irritation occurs seek medical advice.
Eye	If in eyes, hold eyes open, flood with water for at least 15 minutes and see a doctor.
First Aid Facilities	Eye wash station and normal washroom facilities.
Advice to Doctor	Product is a solution of sodium hypochlorite. If swallowed, may lead to fall in blood pressure. Treat with antacids to neutralise hypochlorous acid formed in the stomach, then as for alkaline materials. Onset of pulmonary oedema, following inhalation overexposure, may be delayed. Treat symptomatically. Contact Poisons Information Centre.

Section 5 - Firefighting Measures

Suitable Extinguishing Media	Use extinguishing media appropriate to surrounding fire.
Hazards from Combustion Products	Chlorine, water vapour, sodium hydroxide, sodium carbonate, sodium chloride.
Specific Methods	In case of small fire/explosion use water. In case of major emergency use PPE: breathing apparatus and protective gloves.
Specific Hazards Arising from the Chemical	May form explosive products with primary aliphatic or aromatic amines, methanol and with nitrites. Contact with acids will generate chlorine, a toxic and corrosive gas. May react vigorously or violently with oxidising agents, reducing agents and metal salts.
Other Information	Avoid contact with coloured fabric as Chlorine may bleach colour out. May give off dangerous gas if mixed with other products.

Section 6 - Accidental Release Measures

Spills & Disposal	Disposal of small spillages only. For large spillages liquids should be contained using sand or earth, and both liquids and solids then transferred to salvage containers. Residues should be treated as for small spillages. CAUTION: Before dealing with spillage take necessary protective measures, inform others to keep at a safe distance and, for flammable materials, shut off all possible sources of ignition. If local regulations permit, mop up with plenty of water and run to waste, diluting greatly with running water. Otherwise absorb on inert absorbent, transfer to sealed container and arrange removal by disposals company. Wash site of spillage thoroughly with water and detergent. Ventilate area to dispel any residual vapours.
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Section 7 - Handling and Storage

Conditions for safe storage, including any incompatibilities	Store in a cool, well ventilated place, out of reach of children. Large quantities should be stored in a bunded dangerous goods store. Store in original container. Never store in unlined metal containers. Keep container tightly closed. Keep out of direct sunlight. Keep away from combustible materials. Keep away from acids. Keep away from metals and their salts. Keep away from aliphatic and aromatic amines. Keep away from methanol and nitrites. Keep away from oxidising and reducing agents. Prevent vapours from collecting in enclosed spaces. Protect from physical damage. Clean up all spills and splashes promptly; avoid secondary accidents.
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Section 8 - Exposure Controls and Personal Protection

Engineering Controls	Prevent direct contact with metals. Local mechanical exhaust/extraction usually required to keep airborne contamination as low as possible.
Personal Protective Equipment	Prevent contact with the eyes. Avoid contact with the skin. Avoid breathing the vapours. Personal protection to be selected from those recommended below, as appropriate to mode of use, quantity handled and degree of hazard:- Safety glasses Gloves, rubber or plastic Always maintain a high level of personal hygiene when using cleaning chemicals. That is wash hands before eating, drinking, smoking or using the toilet.

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Section 9 - Physical and Chemical Properties

Form	Liquid
Appearance	Clear slightly yellow liquid
Odour	Smell of lemon and chlorine.
Boiling Point	100C
Solubility in Water	Miscible with water in all proportions.
Specific Gravity	1.1
pH	12.6 - 13.2
Vapour Pressure	Not available.
Flash Point	None
Flammability	Not flammable. Moderate oxidiser.
Other Information	Oxidiser. Contact with combustible material may cause fire. Contact with acids will generate chlorine, a toxic and corrosive gas. May react violently with reducing agents. Can react with primary aliphatic and aromatic amines, methanol and nitrites to give explosive products. May react vigorously with oxidising agents. Incompatible with most metals. Will decompose on standing, generating chlorine. Decomposition will be accelerated by contamination and by exposure to sun light. May react vigorously with peroxides and metal salts. On long storage, may generate pressure inside sealed containers. Open cautiously.

Section 10 - Stability and Reactivity

Possibility of Hazardous Reactions	My form toxic oxides of Chlorine if involved in a fire.
Conditions to Avoid	Heat, flames, ignition sources and incompatibles.
Incompatible Materials	Acids, oxidizing agents, metals
Hazardous Decomposition Products	Decomposes on heating to emit toxic fumes. Heating can cause expansion of containers or decomposition leading to violent rupture of containers. Reacts vigorously with acids to produce dangerous levels of gaseous chlorine. Reacts with amines, ammonium salts, aziridine, methanol, phenylacetonitrile, metal salts, peroxides and reducing agents.

Section 11 - Toxicological Information

Acute Toxicity - Oral	LD50: Sodium Hypochlorite 8910 mg/kg oral, rat
Acute Toxicity - Inhalation	LC 50 : Chlorine 293 ppm/1 hour rat LC50 : Chlorine 2,530 mg/m3/30 minutes human
Ingestion	Will cause irritation and corrosion of the mouth, throat and gastrointestinal system. May cause pain and vomiting.
Inhalation	Inhalation of chlorine gas at 1 ppm will irritate the mouth, nose and throat. Above 1.3 ppm, vapours may cause coughing and difficulty breathing. Risk of delayed onset of pulmonary oedema (fluid in the lungs).
Skin	Short contact may cause irritation. On longer contact risk of chemical burns.
Eye	Severe irritant. Risk of permanent damage.
Chronic Effects	Repeated skin contact may lead to dermatitis or 'chloracne'. Repeated, low level exposure to chlorine vapours may cause corrosion of the teeth.

Section 12 - Ecological Information

Ecotoxicity	Toxic to fish and aquatic organisms.
Mobility	Readily dilutes with water.
Information on Ecological Effects	Harmful to aquatic life.

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Environmental Protection Avoid contaminating waterways, drains, sewers, or ground.

Section 13 - Disposal Considerations

Waste Disposal Land fill, sewer (small quantities). Refer to Land Waste Management Authority in your State.

Section 14 - Transport Information

Transport Information Not classified as a Dangerous Good according to the Australian Code for the Transport of Dangerous Goods by Road and Rail. Store away from acids.

Section 15 - Regulatory Information

Poisons Schedule S5

Australia (AICS/AIIC) All components listed.

Section 16 - Any Other Relevant Information

Date of Preparation 3/11/2024

Literature References Preparation of Safety Data Sheets for hazardous Chemicals Code of Practice Standard for the Uniform Scheduling of Medicines and Poisons Australian Code for the Transport of Dangerous Goods by Road & Rail Globally Harmonised System of classification and labelling of chemicals GHS7

Signature of Preparer/Data Service Technical manager Tel: (03) 9450 4555

Technical Contact Numbers Emergency Advice All Hours:
Chief Chemist Tel: (03) 9450 4555 Mon-Fri 8am - 6pm
Poisons Information Centre: 13 11 26 - 24hrs

Other Information This SDS summarises at the date of issue our best knowledge of the health and safety hazard information of the product, and in particular how to safely handle and use the product in the Workplace. Please refer to the technical datasheet (Instructions for use), and the label on the drum. The company cannot anticipate or control the individual working conditions encountered and so each user should read this SDS carefully, and if in doubt ring the Contact Point Number given below.
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